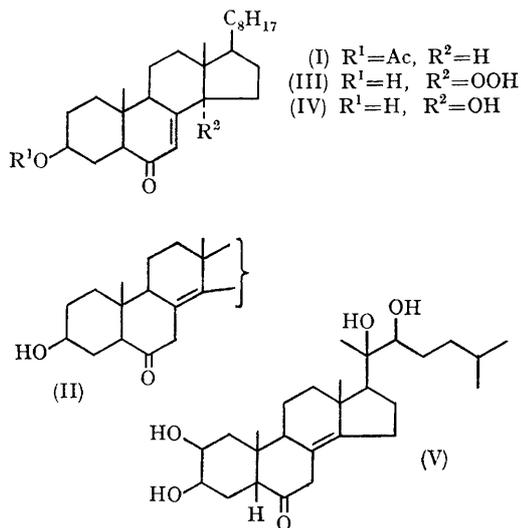


Photo-oxygenation of $\beta\gamma$ -Unsaturated Ketones

By N. FURUTACHI, Y. NAKADAIWA, and K. NAKANISHI*

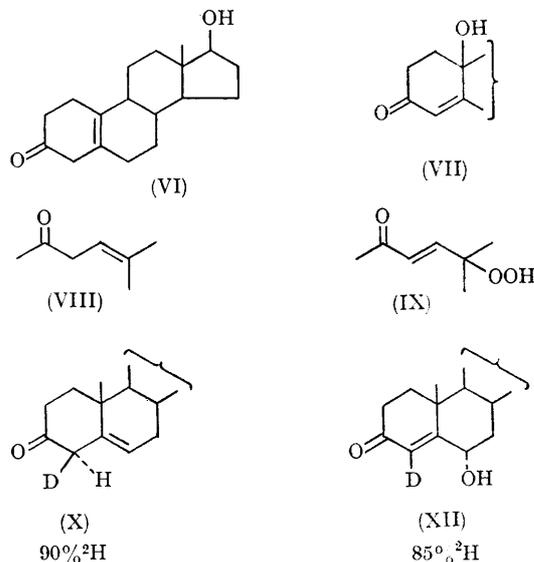
(Department of Chemistry, Tohoku University, Sendai, Japan)

THE mechanism of photosensitized oxygenation of isolated olefins has been greatly clarified in recent years.¹ Participation of the excited singlet-state oxygen is now generally accepted,² and a concerted cycloaddition of oxygen to olefins containing allylic hydrogens is favoured.³ We have investigated the photosensitized oxidations of $\beta\gamma$ -unsaturated ketones, and preliminary data indicate that also in these systems, which contain more labile allylic hydrogens, a concerted cycloaddition is favoured over a step-wise ionic addition.⁴

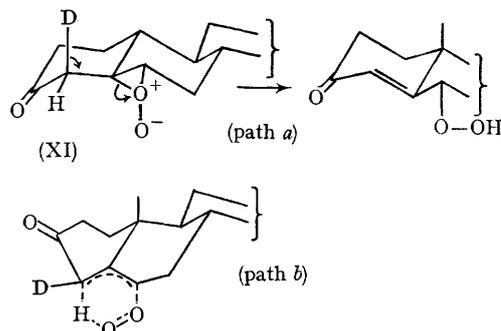


3β -Acetoxycholest-7-en-6-one (I)⁵ was converted into (II) upon brief treatment with Bu^tOK-Bu^tOH ⁶ followed by rapid neutralization with acetic acid. Irradiation of (II) in pyridine with a 250 w tungsten lamp in the presence of Rose Bengal under oxygen gave the hydroperoxide (III), which upon reduction with $NaI-AcOH$ yielded the authentic 14α -hydroxy-enone (IV),⁵ the yield of (IV) from (I) being *ca.* 70%. Similarly, the $\beta\gamma$ -unsaturated ketone (V) yielded ponasterone A⁷ (14α -hydroxy-7-en-6-one) *via* the hydroperoxide in almost quantitative yield, and hence this photo-oxygenation provides an alternative method⁸ for making the 14α -hydroxy-7-en-6-one moiety present in all ecdysones discovered to date (about twenty).

Photosensitized oxidation of 17β -hydroxy-19-norandrost-5,10-en-3-one (VI), prepared from oestradiol⁹ by Birch reduction, was also transformed into 10β -hydroxy-19-nortestosterone (VII)⁹ *via* the hydroperoxide in 45% overall yield. Likewise, 5-methylhex-4-en-2-one (VIII)¹⁰ was quantitatively oxidized to the hydroperoxide (IX).†



The cyclic nature of olefin photo-oxygenation has been elegantly proven with $[7-^2H]$ cholesterol molecules.^{3a} $\beta\gamma$ -Unsaturated ketones, in which the hydrogen(s) flanked by the carbonyl group and



† Structure proven by spectroscopic data.

double bond dissociate readily, are suitable systems for studying the participation of an ionic step-wise mechanism. 4 β -Deuteriocholest-5-ene (X)¹¹ (90% deuterium content according to mass and n.m.r. spectroscopy) was therefore synthesized and submitted to the photosensitized oxygenation. If a step-wise mechanism going through an intermediate such as (XI) (or its equivalent) were operating, most of the deuterium should be lost (path *a*). In contrast, path *b* should retain the

deuterium. The hydroperoxide was reduced to 6 α -hydroxycholest-4-en-3-one (XII), and the deuterium content at C-4 was measured from the n.m.r. curve, taking the 6 β -proton as reference for integration. This showed that the product (XII) contained 85% deuterium at C-4, and hence it is very likely that the photo-oxygenation of $\beta\gamma$ -unsaturated ketones is a concerted cycloaddition.

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¹ For leading references, see K. Gollnick and G. O. Schenck, in "1,4-Cycloaddition Reactions; the Diels-Alder Reaction in Heterocyclic Syntheses", ed. J. Hamer, Academic Press, New York, 1967, ch. X, p. 255; see also K. Gollnick, Internat. Oxid. Symp., San Francisco, Abstract II, 1967, p. 383; A. U. Khan and D. R. Kearns, *ibid.*, p. 633; R. Higgins, C. S. Foote, and H. Cheng, *ibid.*, p. 672.

² C. S. Foote and S. Wexler, *J. Amer. Chem. Soc.*, 1964, **86**, 3879, 3880; E. J. Corey and W. C. Taylor, *ibid.*, p. 3881; D. R. Kearns, R. A. Hallins, A. U. Kahn, R. W. Chamber, and P. Radlick, *ibid.*, 1967, **89**, 5455.

³ (a) A. Nickon and J. F. Bagli, *J. Amer. Chem. Soc.*, 1959, **81**, 6330; 1961, **83**, 1498; (b) F. A. Litt and A. Nickson, Internat. Oxid. Symp., San Francisco, Abstract II, 1967, p. 613.

⁴ K. F. Kopecky and H. J. Reich, *Canad. J. Chem.*, 1965, **43**, 2265.

⁵ We are grateful to Dr. T. Wada, Teikoku Hormone Company, for the generous gifts of (I), (VI), and authentic (IV).

⁶ H. J. Ringold and S. K. Malhotra, *Tetrahedron Letters*, 1962, 669.

⁷ K. Nakanishi, M. Koreeda, S. Sasaki, M. L. Chang, and H. Y. Hsu, *Chem. Comm.*, 1966, 915.

⁸ A. Furlenmeier, A. Furst, A. Langemann, G. Waldvogel, U. Kerb, P. Hocks, and R. Wiechert, *Helv. Chim. Acta*, 1966, **49**, 1591; J. B. Siddall, J. P. Marshall, A. Bowers, A. D. Cross, J. A. Edwards, and J. H. Fried, *J. Amer. Chem. Soc.*, 1966, **88**, 379; H. Mori, K. Shibata, K. Tsuneda, M. Sawai, *Chem. and Pharm. Bull. (Japan)*, 1968, **16**, 563.

⁹ J. P. Ruelas, J. Iriarte, F. Kinch, and C. Djerassi, *J. Org. Chem.*, 1958, **23**, 1744. A similar reaction has been carried out with 17 α -ethynyl-17 β -hydroxyoestr-5(10)-en-3-one: T. Legatt and E. L. Shapiro, U.S.P. 3,280,157 (*Chem. Abs.*, 1967, **66**, 29,000). We are grateful to the referee for this information.

¹⁰ N. C. Yang and M. J. Jorgenson, *Tetrahedron Letters*, 1964, 1903.

¹¹ S. K. Malhotra and H. J. Ringold, *J. Amer. Chem. Soc.*, 1965, **87**, 3228.